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Hydrogen gas was cooled from 150 oc to 50 oc. Its new pressure is 75.0 torr. What was its original pressure? 323k Chlorine gas occupies a volume of 25.0 ml, at 300 K. What volume will it occupy at 600 K? 25.0m L (000k $X = 60$ 300 A sample of neon gas at 50 oc and a pressure of 2.50 mmHg is cooled to 25 oc. What is the new pressure? X —

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4.two liquids wich can mix are said to be __ (8 Letters) 13.moles of a solute per kilogram of solvent (8 letters) 14.solution containing a relatively large amount of solvent (6 letters) 19.these substances dissociate or ionize in water and are then able to conduct an electric current (12 letters) 5.solution containing a relatively large amount of dissolved solute (12 letters) 8.the solubility ...

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